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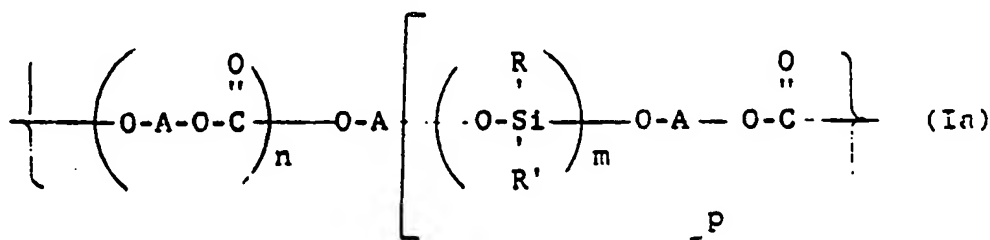
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(54) Thermoplastic polycarbonates and blends with enhanced flame retardant properties.

(57) It has now been found that the flame retardance of polydiorganosiloxane copolycarbonates may be significantly improved upon the addition thereto of a small amount of a metal salt of an inorganic acid. The polydiorganosiloxane copolycarbonates are conforming structurally to



wherein

p is an integer of from 1 to about 10,

n is an integer of from 1 to about 1000,

m is an integer of from 1 to about 500,

A is a derivative of an aromatic dihydroxy compound,

R and R' independently denote a C<sub>1</sub>-C<sub>20</sub> alkyl radical or a C<sub>6</sub>-C<sub>14</sub> aryl radical,

which copolymer is characterized in that its relative viscosity measured on a solution of 0.5 g in 100 ml of dichloromethane at 25° C is at least 1.15.

EP 0 322 690 A1

**THERMOPLASTIC POLYCARBONATES AND BLENDS WITH ENHANCED FLAME RETARDANT PROPERTIES**Field of the Invention

The invention is directed to a thermoplastic polycarbonate composition and more particularly to a flame  
 5 retardant copolycarbonate containing siloxane structured units in its molecular chains.

Background of the Invention

10 Aromatic polycarbonate resins are well known and are accepted as suitable for a wide variety of uses including injection molding, extrusion and film formation. The chemistry, synthesis, properties, and applications of these polycarbonates are extensively discussed in Chemistry and Physics of Polycarbonates by Schnell, Interscience, 1964 and Polycarbonates by Christopher and Fox, Reinhold, 1962.

Although polycarbonates have some inherent flame resistance, being self-extinguishing, ever more  
 15 demanding flame retardancy requirements have spawned numerous attempts to increase this property. Among the methods to improve the flame retardance there is noted adding substantial amounts of halogen to polycarbonate resins such as by employing a halogen substituent on the polycarbonate polymer chain (U.S. Patent 3,751,400 and 3,334,154) or adding a monomeric halogenated compound as in U.S. Patent 3,382,207. Also noted, is the approach taught by U.S. Patent 3,775,367 which entails adding an organic  
 20 and/or inorganic metal salt to the composition. Additions of perfluoroalkane sulfonic acid salts of alkali metals and organic alkali salts of carboxylic acids have been disclosed in U.S. Patent 3,836,490 and in U.S. Patent 3,775,367. Also noted is German Published Patent No. 2,149,311 which discloses the use of insoluble alkali metal salts, particularly salts of inorganic acids, phosphonic acids and sulphonic acids. Flame retardant polycarbonate compositions containing alkali metal salts of an inorganic acid are disclosed  
 25 in U.S. Patent 4,223,100. U.S. Patent 3,535,300 discloses the use of small amounts of specified metal salts (which do not include alkali metal salts) in combination with halogen carried on the polymeric backbone or on an additive. U.S. Patent 4,110,299 teaches adding of alkali or alkaline earth metal salts of organic acids in combination with additives such as inorganic halide and an organic monomeric or polymeric aromatic or heterocyclic halide improve the flame retardancy of an aromatic polycarbonate.

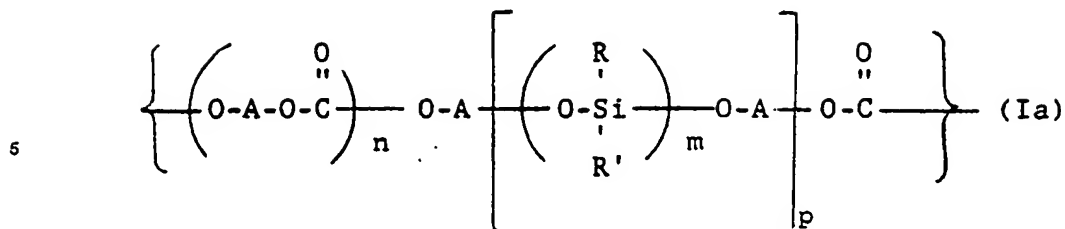
30 Also noted are the teaching in regard to the drip suppression of the compositions when they are exposed to a flame. In German Patent No. 2,535,262 there is a teaching to add fluorinated polyolefins such as polytetrafluoroethylene (PTFE) to a polycarbonate containing organic alkali metal salt to retard dripping. U.S. Patent 4,110,299 discloses adding to a polycarbonate resin fluorinated polyolefin, fibrous glass or a siloxane, in combination with certain salts and an inorganic halide to diminish tendency to drip. In PCT  
 35 application WO 80/00084 there is disclosed - in Example 13 - a blend of aromatic polycarbonate and a block copolymer of polycarbonate and polydimethylsiloxane (57%/43%) and including sodium salt of trichloro benzene sulfonic acid. The blend was noted to have improved resistance to stress crazing. The document generally discloses improved ductility, solvent resistance and flame retardance characteristics of blends of polycarbonates and a block copolymer of polycarbonate and polydiorganosiloxane.

40 Polydiorganosiloxane copolycarbonates of the type entailed in the present invention are known.

SUMMARY OF THE INVENTION

45 It has now been found that the flame retardance of polydiorganosiloxane copolycarbonates may be significantly improved upon the addition thereto of a small amount of a metal salt of an inorganic acid. The polydiorganosiloxane copolycarbonates conforming structurally to

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wherein

n is an integer of from 1 to about 1000,

m is an integer of from 1 to about 500,

p is an integer of 1 to 10,

A is a derivative of an aromatic dihydroxy compound,

P and R' independently denote a C<sub>1</sub>-C<sub>20</sub> alkyl radical or a C<sub>6</sub>-C<sub>14</sub> aryl radical,

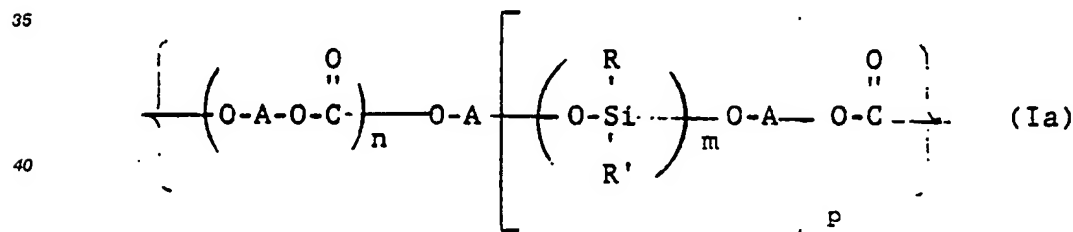
which copolymer is characterized in that its relative viscosity measured on a solution of 0.5 g in 100 ml of dichloromethane at 25° C is at least 1.15.

### DETAILED DESCRIPTION OF THE INVENTION

The suitable siloxane copolycarbonates are characterized in that the relative viscosity, measured on a solution of 0.5 g in 100 ml of dichloromethane at 25° C, is at least 1.15 preferably from about 1.18 to about 1.38 and contain about 0.5 to about 60% preferably about 1.0 to about 15% by wt. of polydiorganosiloxane segments. These polydiorganosiloxane segments contain an average of about 5 to about 200 preferably 7 to about 160 silicone atoms per segment. A segment in the present context is a siloxane chain which is not interrupted by a carbon atom.

The polydiorganosiloxane copolycarbonates of the invention are well known in the art. U.S. Patents 3,189,662; 3,419,634; 3,821,325; 3,832,419 and 4,584,360 all contain descriptions of the copolycarbonate and some disclose the method for preparation. These patents are incorporated herein by reference.

The structure of the copolycarbonate of the invention is



wherein

n is an integer of from 1 to about 1000,

m is an integer of from 1 to about 500,

p is an integer of from 1 to about 10,

A is a derivative of an aromatic dihydroxy compound and

R and R' independently are a C<sub>1</sub>-C<sub>20</sub> alkyl or a C<sub>6</sub>-C<sub>14</sub> aryl radical. Both the aryl and alkyl radicals may be substituted by halogen atoms.

Most preferred R and R' radicals are selected from the group consisting of methyl, ethyl, propyl, n- and tert.-butyl, chloromethyl, trifluoropropyl, phenyl, chlorophenyl and naphthyl. The block copolymers of the invention comprise about 40 to 99.5, preferably 85 to 99 percent by weight of polycarbonate segments.

The preferred derivatives of diphenol radicals are:



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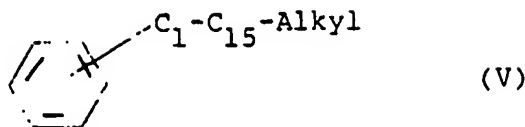
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In which

Z denotes -O- or -  $\text{C} = \text{O}$  -

Examples of preferred terminal groups are the radicals of phenol, p-tert.-butyl-phenol, p-tert.-octyl-phenol, p-nonyl-phenyl, 3,5-di-tert.-butyl-phenol, 2-(3,5-dimethylheptyl)-phenol and 4-(3,5-dimethylheptyl)-phenol.

In place of free phenols, their halocarboxates may be employed, and in place of the carboxylic acids, their acid chlorides may be employed.

The amount of terminal groups is preferably 1 to 10 mol %, relative to the aromatic dihydroxy component.

The block copolymer may be prepared by the two-phase interfacial process under conventional conditions from , -bis(hydroxyaryloxy)-polydiorganosiloxanes, bisphenols, carbonic acid derivatives, for example phosgene, optionally with chain terminators and/or branching agents. The transesterification process and the process in a homogenous solution are also suitable for the preparation.

The salts useful in the present invention include those formed between alkali metal or alkaline earth metals and inorganic acids. Lithium, sodium, potassium, magnesium, calcium and barium salts are preferred. Inorganic acids in the present context include any compound which meets the traditional tests of acidity and contains a complex ion. Preferred acids contain complex oxo-anions or fluoro-anions.

The most preferred are the alkali or alkaline-earth metal complex fluoro anion salts or complex oxo-anion salts. This terminology being derived from the discussion of fluorine compounds contained in the text "Advanced Inorganic Chemistry" by F.A. Cotton and G. Wilkinson, Interscience Publishers, 1962, incorporated herein by reference. Suitable inorganic alkali metal complex fluoro anion salts include  $\text{KBF}_4$ ,  $\text{K}_3\text{AlF}_6$ ,  $\text{KAlF}_4$ ,  $\text{K}_2\text{SiF}_6$ ,  $\text{Na}_3\text{AlF}_6$ ,  $\text{KPF}_6$ ,  $\text{NaSbF}_6$ ,  $\text{Na}_3\text{FeF}_6$ ,  $\text{NaPF}_6$ ,  $\text{Na}_2\text{SiF}_6$ ,  $\text{Na}_2\text{TiF}_6$ ,  $\text{NaBF}_4$ ,  $\text{K}_2\text{TaF}_7$ ,  $\text{K}_2\text{NbF}_7$ ,  $\text{KSbF}_6$ ,  $\text{K}_2\text{NiF}_6$ ,  $\text{K}_2\text{TiF}_6$ ,  $\text{LiBF}_4$ ,  $\text{LiPF}_6$ ,  $\text{Li}_2\text{BeF}_4$ ,  $\text{Li}_3\text{AlF}_6$ ,  $\text{MgSiF}_6$ , and  $\text{BaSiF}_6$ .

$\text{Li}_3\text{AlF}_6$ ,  $\text{BaSiF}_6$ ,  $\text{KBF}_4$ ,  $\text{K}_3\text{AlF}_6$ ,  $\text{KAlF}_4$ ,  $\text{K}_2\text{SiF}_6$  and  $\text{Na}_3\text{AlF}_6$  are the preferred metal complex fluoro anion salts, and  $\text{Na}_3\text{AlF}_6$  (cryolite) is the most preferred complex fluoro anion salt. Among the suitable complex oxo-anion salts are the alkali and alkaline-earth salts of carbonic acid, such as  $\text{Na}_2\text{CO}_3$ ,  $\text{K}_2\text{CO}_3$ ,  $\text{MgCO}_3$ ,  $\text{CaCO}_3$ ,  $\text{BaCO}_3$  and  $\text{BaVO}_3$ . Other suitable salts are sulfates, phosphates and tungstenates. The most preferred salts of oxo-anion complex acids are  $\text{BaCO}_3$  and  $\text{BaVO}_3$ .

The salts may be used in any effective amount up to about 2 wt. % based on the total weight of the copolymer. It is preferred to use no less than about 0.05 wt. % and more preferably no less than about 0.1 wt. %. It is also preferred to use no more than about 1.5% and more preferably no more than 0.8%.

The compositions of the invention may contain conventional additives such as pigments, dyes, UV stabilizers, thermal stabilizers, mold release agents, reinforcing agents and fillers. The compositions may be formed by any method commonly known in the art. For instance, the various additives may be dry blended with the copolymer pellets and the mixture extruded. In the context of the invention, the copolycarbonates may be blended with a thermoplastic polycarbonate which is essentially siloxane-free. The siloxane content in this blend is preferably between about 0.5 to about 15% by wt.

The invention is further illustrated, but is not intended to be limited by the following examples in which all parts and percentages are by weight unless otherwise specified.

## EXAMPLES

### Example 1

Several compositions were prepared and their properties evaluated as follows. In the compositions numbered 1-4 - see the following table. Resin A was a polydimethylsiloxane block copolymer containing 5%

by weight of siloxane segments having average segment length of 80 and characterized in that its relative viscosity is 1.26-1.27.

	1	2	3	4
Resin A	99.6	99.2	98.5	100
Cryolite	0.4	--	--	--
Oligotetrabromo bisphenol A carbonate	--	0.8	1.5	--
Relative Viscosity	1.257	1.271	1.259	1.267
MFR <sup>(1)</sup> , gms/10 min.	5.80	5.58	5.88	4.96
Impact Strength Notched Izod				
1/8 in., ft. lbs/in.	14.3	15.2	14.7	14.9
1/4 in., ft. lbs/in.	12.2	12.2	12.2	12.4
UL-94 Flame Test				
Unaged 1/8 in. Class	V-0	V-0	V-0	V-1
Burn time (2)	2.1	2.1	2.1	5.6
Aged 7 days 1/8 in. Class	V-0	V-0	V-0	--
Burn time	2.4	2.1	1.9	--
Unaged 1/16 in. Class	V-0	V-2	Fails V-2	V-2
Burn time	2.5	--	--	--
Aged 7 days 1/16 in. Class	V-0	(3)	(3)	(3)
Burn time	3.3			

(1) Melt flow rate in accordance with ASTM D-1238.

(2) Average burning time, seconds.

(3) Failed V-2.

The performance of compositions containing cryolite (composition 1), an oligomeric tetrabromo carbonate (Compositions 2 and 3), was compared to that of the unmodified resin.

As can be seen, the flammability of the resin by itself or when it incorporates known flame retarding agents is inferior to that which is demonstrated by the composition in accordance with the invention. This performance has not been reported or suggested by the prior art.

## Example 2

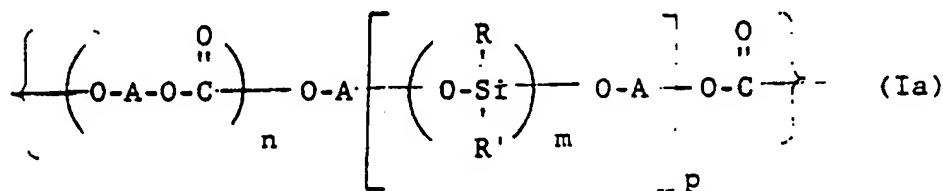
In the Example below there is a direct comparison between Resin A and Makrolon 2508 which is a homopolycarbonate based on bisphenol A, having a relative viscosity of about 1.26. In both compositions there was incorporated 0.40% of cryolite. As the example demonstrates the composition of the invention is superior to the composition based on Makrolon 2508 in term of its flammability rating.

	7	8
Resin A	--	99.60
MAKROLON 2508, %	99.60	--
CRYOLITE	0.40	0.40
RELATIVE VISCOSITY	1.256	1.258
MFR, gms/10 min.	15.70	6.3
NOTCHED IZOD IMPACT		
1/8 in., ft lbs/in.	15.07	14.90
1/4 in., ft lbs/in.	1.81	13.27
CRIT THICKNESS, mils	135	>470
UL-94 FLAME TEST		
1/8 in. CLASS	V-1	V-0
1/16 in. CLASS	V-2	V-0

Although the Invention has been described in detail in the foregoing for the purpose of illustration, it is to be understood that such detail is solely for that purpose and that variations can be made therein by those skilled in the art without departing from the spirit and scope of the invention except as it may be limited by the claims.

## Claims

1. A thermoplastic molding composition comprising (i) a polydiorganosiloxane-polycarbonate block copolymer conforming structurally to



wherein

p is an integer of from 1 to about 10,

n is an integer of from 1 to about 1000,

m is an integer of from 1 to about 500,

A is a derivative of an aromatic dihydroxy compound,

R and R' independently denote a C<sub>1</sub>-C<sub>20</sub> alkyl radical or a C<sub>6</sub>-C<sub>14</sub> aryl radical,

which copolymer is characterized in that its relative viscosity measured on a solution of 0.5 g in 100 ml of dichloromethane at 25 °C is at least 1.15 and (ii) an amount of metal salt of an inorganic acid, said amount being sufficient for improving the flammability rating of said copolymer.

2. The composition of Claim 1 wherein said salts are metal complex fluoro anion salts or metal complex oxo-anion salts.

3. The composition of Claim 1 wherein said amount is about 0.05 to about 2% by wt. relative to the weight of the composition.

4. The composition of Claim 2 wherein said salt is selected from the group consisting of KBF<sub>4</sub>, K<sub>3</sub>AlF<sub>6</sub>, KAlF<sub>4</sub>, K<sub>2</sub>SiF<sub>6</sub>, Na<sub>3</sub>AlF<sub>6</sub>, KPF<sub>6</sub>, NaSbF<sub>6</sub>, Na<sub>3</sub>FeF<sub>6</sub>, NaPF<sub>6</sub>, Na<sub>2</sub>SiF<sub>6</sub>, Na<sub>2</sub>TiF<sub>6</sub>, NaBF<sub>4</sub>, K<sub>2</sub>TaF<sub>7</sub>, K<sub>2</sub>NbF<sub>7</sub>, KSbF<sub>6</sub>, K<sub>2</sub>NiF<sub>6</sub>, K<sub>2</sub>TiF<sub>6</sub>, LiBF<sub>4</sub>, LiPF<sub>6</sub>, Li<sub>2</sub>BeF<sub>4</sub>, Li<sub>3</sub>AlF<sub>6</sub>, MgSiF<sub>6</sub>, BaSiF<sub>6</sub>, BaCO<sub>3</sub> and BaVO<sub>3</sub>.

5. The composition of Claim 4 wherein said salt is selected from the group consisting of KBF<sub>4</sub>, K<sub>3</sub>AlF<sub>6</sub>, KAlF<sub>4</sub>, K<sub>2</sub>SiF<sub>6</sub>, Na<sub>3</sub>AlF<sub>6</sub>, Li<sub>3</sub>AlF<sub>6</sub> and BaSiF<sub>6</sub>.

6. The composition of Claim 1 wherein said amount is about 0.1 to about 1.5%.

7. The composition of Claim 1 wherein said amount is about 0.2 to about 0.8%.

8. The composition of Claim 1 wherein said salts are metal complex oxo-anion salts.
9. The composition of Claim 1 wherein said metal salt is selected from the group consisting of  $\text{Na}_2\text{CO}_3$ ,  $\text{K}_2\text{CO}_3$ ,  $\text{MgCO}_3$ ,  $\text{CaCO}_3$ ,  $\text{BaCO}_3$  and  $\text{BaVO}_3$ .
10. The composition of Claim 9 wherein said metal salt is  $\text{BaCO}_3$  or  $\text{BaVO}_3$ .

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# EUROPEAN SEARCH REPORT

Application Number

EP 88 12 1185

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.4)
X	US-A-4 157 997 (L. P. DeMEJO et al.) * Claims *	1,2,8-10	C 08 K 3/00 C 08 L 83/10 C 08 K 3/16
A	US-A-3 742 083 (C.A. BIALOUS)		
D,A	FR-A-2 154 783 (BAYER AG)		
			TECHNICAL FIELDS SEARCHED (Int. Cl.4)
			C 08 K C 08 L
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 06-04-1989	Examiner DE LOS ARCOS Y VELAZQUEZ
<b>CATEGORY OF CITED DOCUMENTS</b>			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	